



Plasmon assisted isomerisation in naphthalene and azulene

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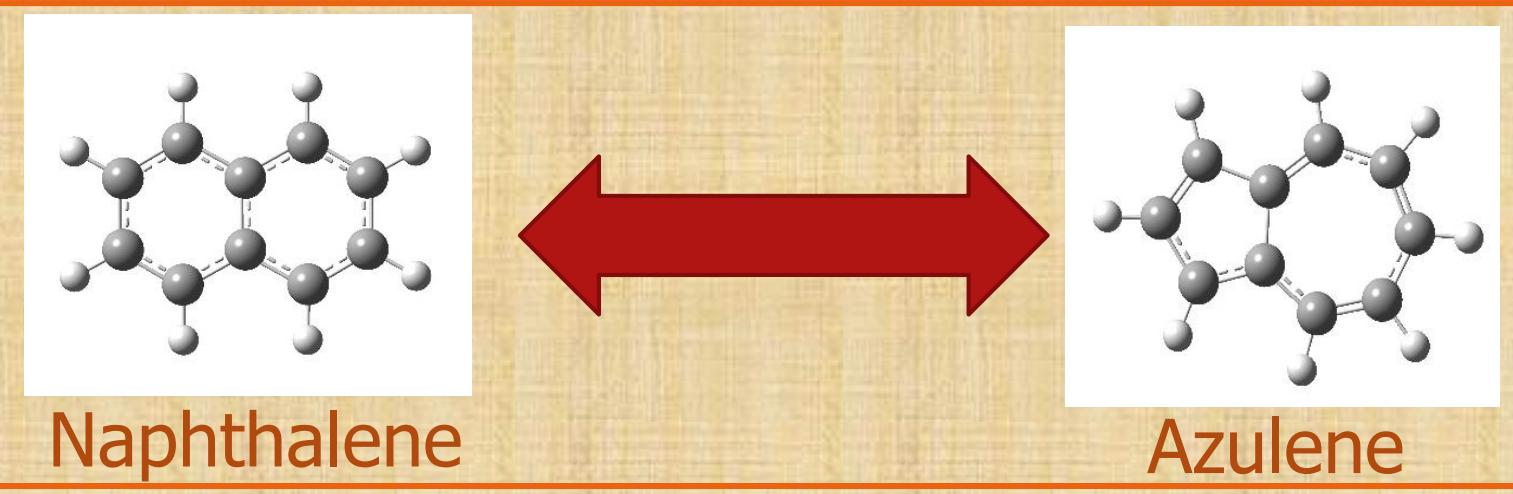
Motivation



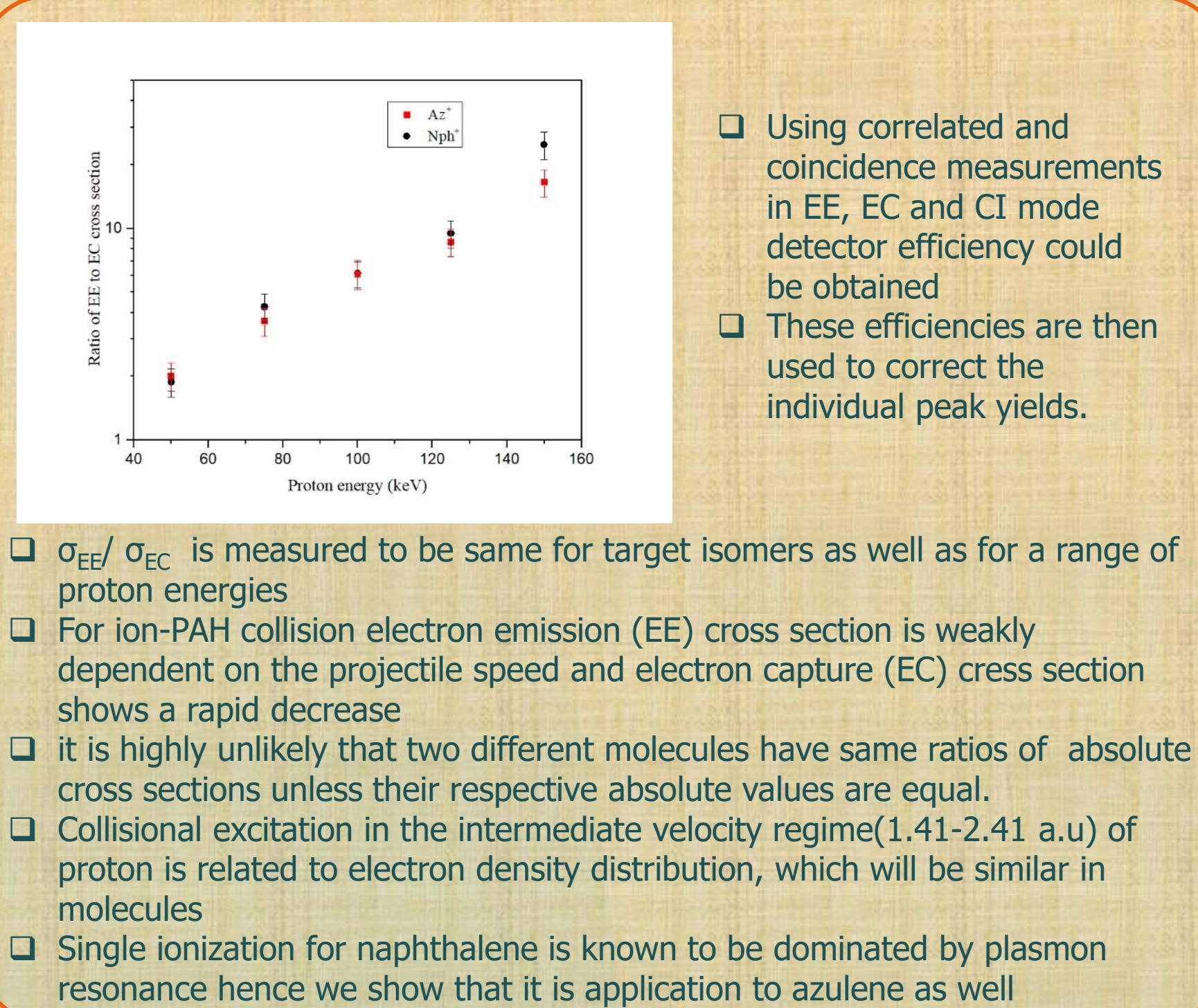
Terrestrial sources of PAHs: Industries and automobiles
http://blog.chemistry-matters.com

- Astrochemistry of ISM is a function of dissociative and isomerization processes in PAHs and their derivatives.
- Molecular dynamics of PAHs is a benchmark for several biologically important processes.
- 6 to 5 membered ring deformation is an important pathway of fullerene formation in ISM.
- Neutral elimination of PAH cations strongly depends on isomerization.
- Azulene and Naphthalene are the simplest isomeric pairs with special attributes.
- Plasmon resonance as a dominant and constant energy loss mechanism in ion-PAH collision.

Isomerisation



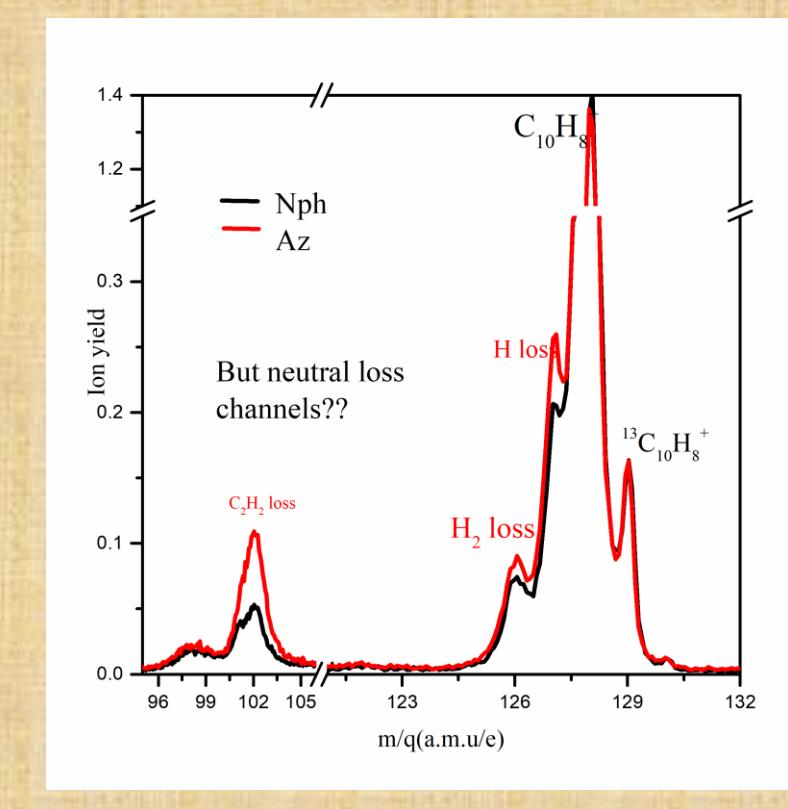
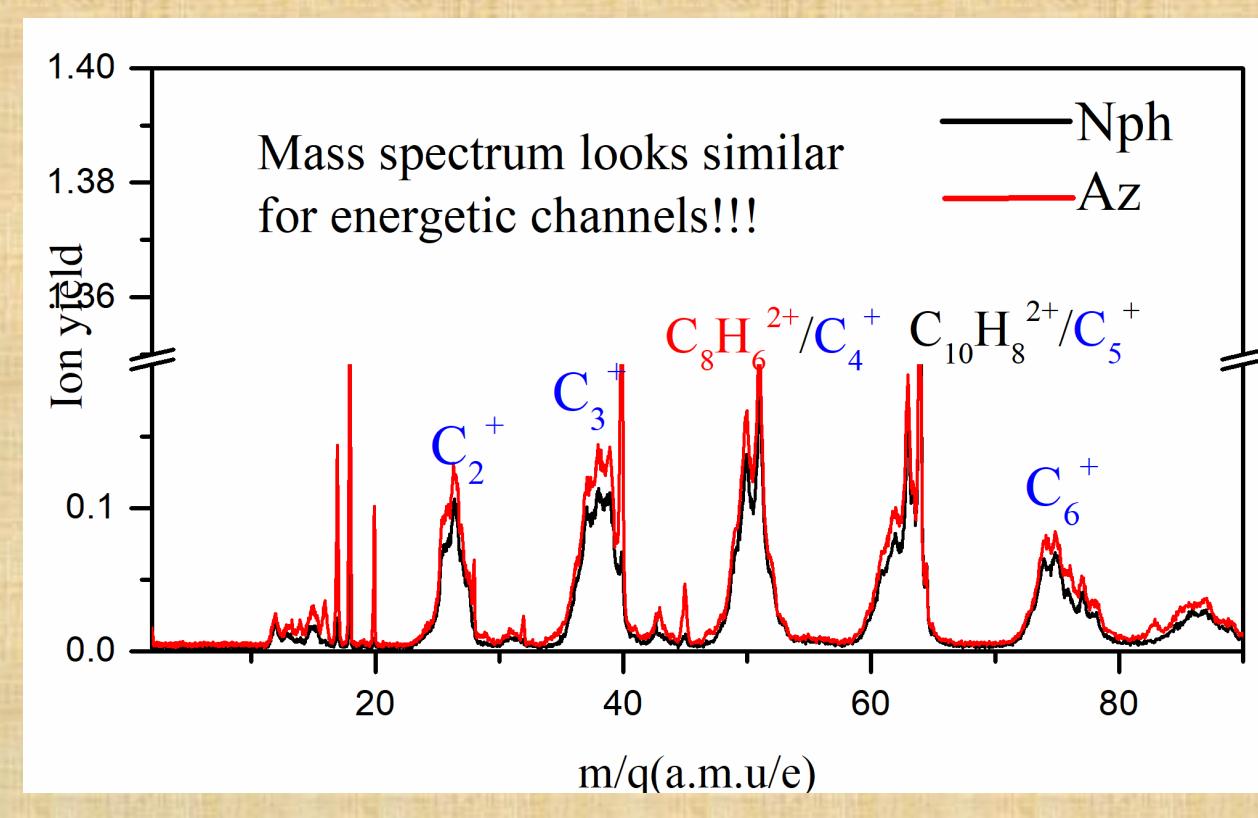
Electron emission & Electron capture cross sections of $C_{10}H_8$ isomers:



- Using correlated and coincidence measurements in EE, EC and CI mode detector efficiency could be obtained
- These efficiencies are then used to correct the individual peak yields.

- σ_{EE}/σ_{EC} is measured to be same for target isomers as well as for a range of proton energies
- For ion-PAH collision electron emission (EE) cross section is weakly dependent on the projectile speed and electron capture (EC) cross section shows a rapid decrease
- it is highly unlikely that two different molecules have same ratios of absolute cross sections unless their respective absolute values are equal.
- Collisional excitation in the intermediate velocity regime (1.41–2.41 a.u.) of proton is related to electron density distribution, which will be similar in molecules
- Single ionization for naphthalene is known to be dominated by plasmon resonance hence we show that it is application to azulene as well

Similarities & differences in the mass spectra of $C_{10}H_8$ isomers → Isomerisation

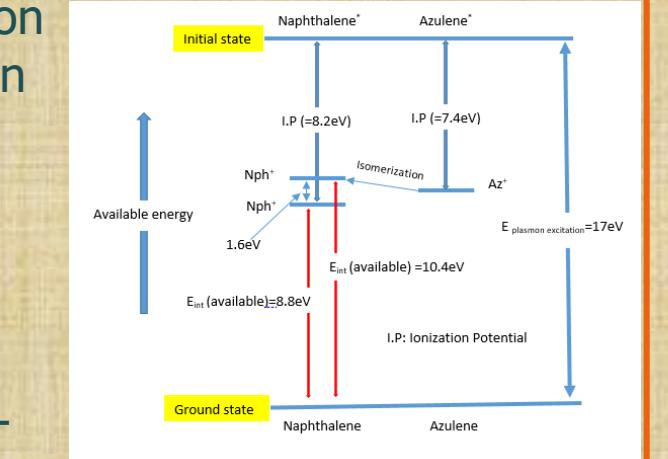


Results & discussions

- Smaller fragments produced via energetic reactions can come from equilibrium mixture of possible parent ions.
- For low energy reactions, isomerisation between cationic az and nph is a steady state reaction, which favours one direction.
- For a fixed energy excitation process (plasmon resonance at 16–17 eV), $k(\text{az}^+) > k(\text{nph}^+)$.
- As per Dyakov et al calculation, Az^+ isomerises to Nph^+ prior to dissociation.
- Internal energy of Nph^+ produced from Az^+ would have enhanced by isomerization and resulted a fast decay, it is conspicuous from the mass spectrum $E_{int}(\text{az}^+) - E_{int}(\text{nph}^+) = E_{iso} + IP_{nph} - IP_{\text{az}} = 1.6 \text{ eV}$

Energy diagram: plasmon resonance and isomerization

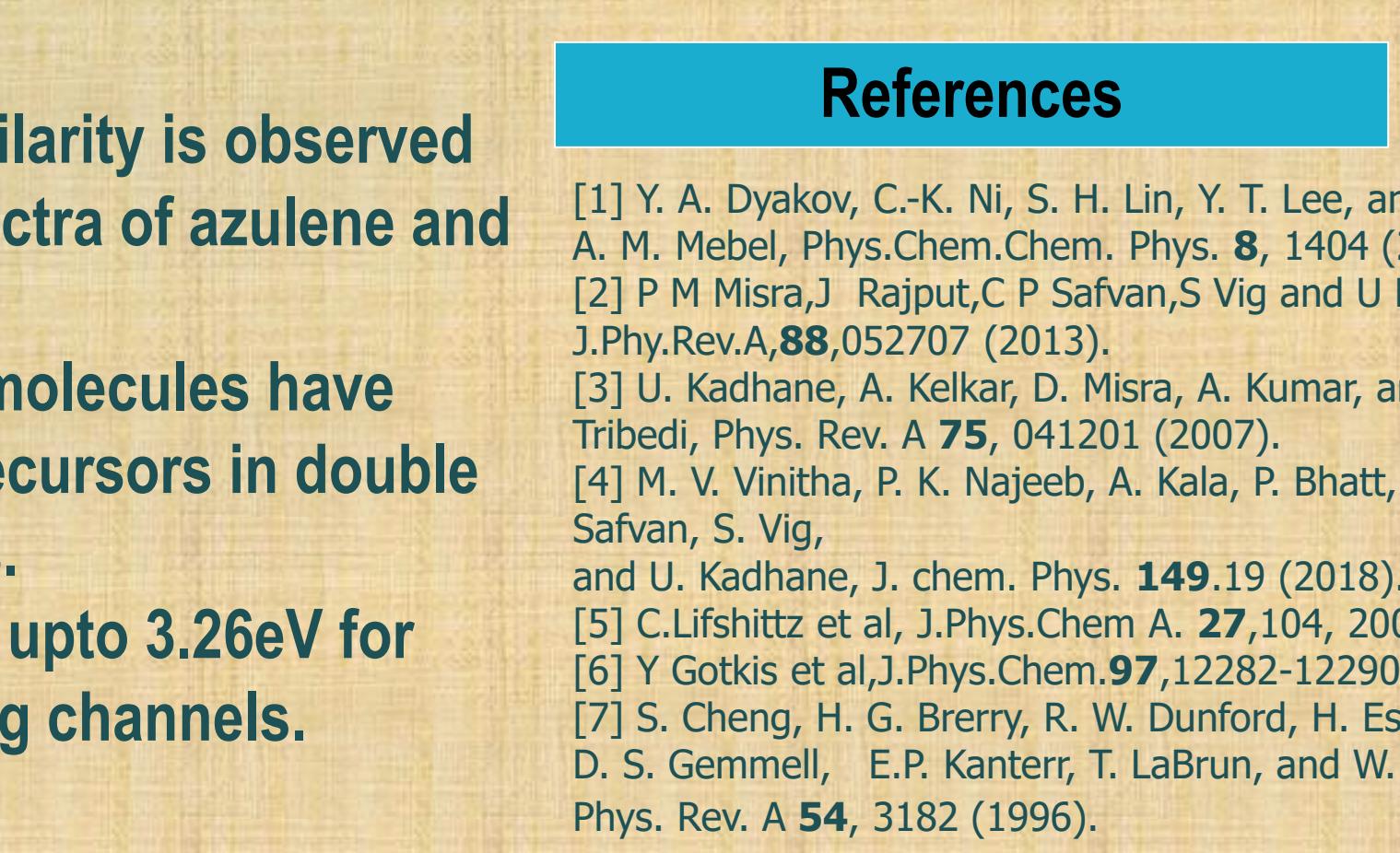
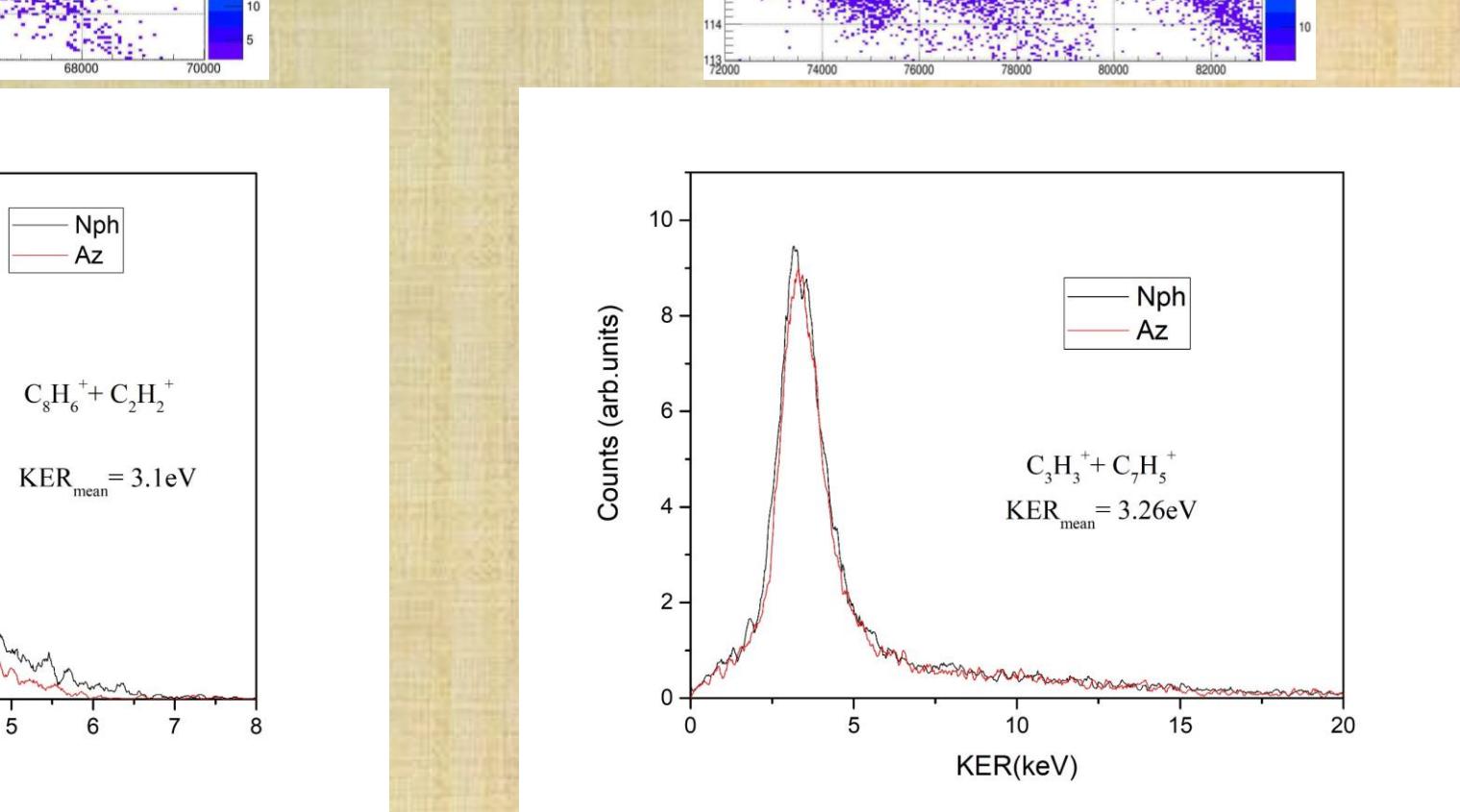
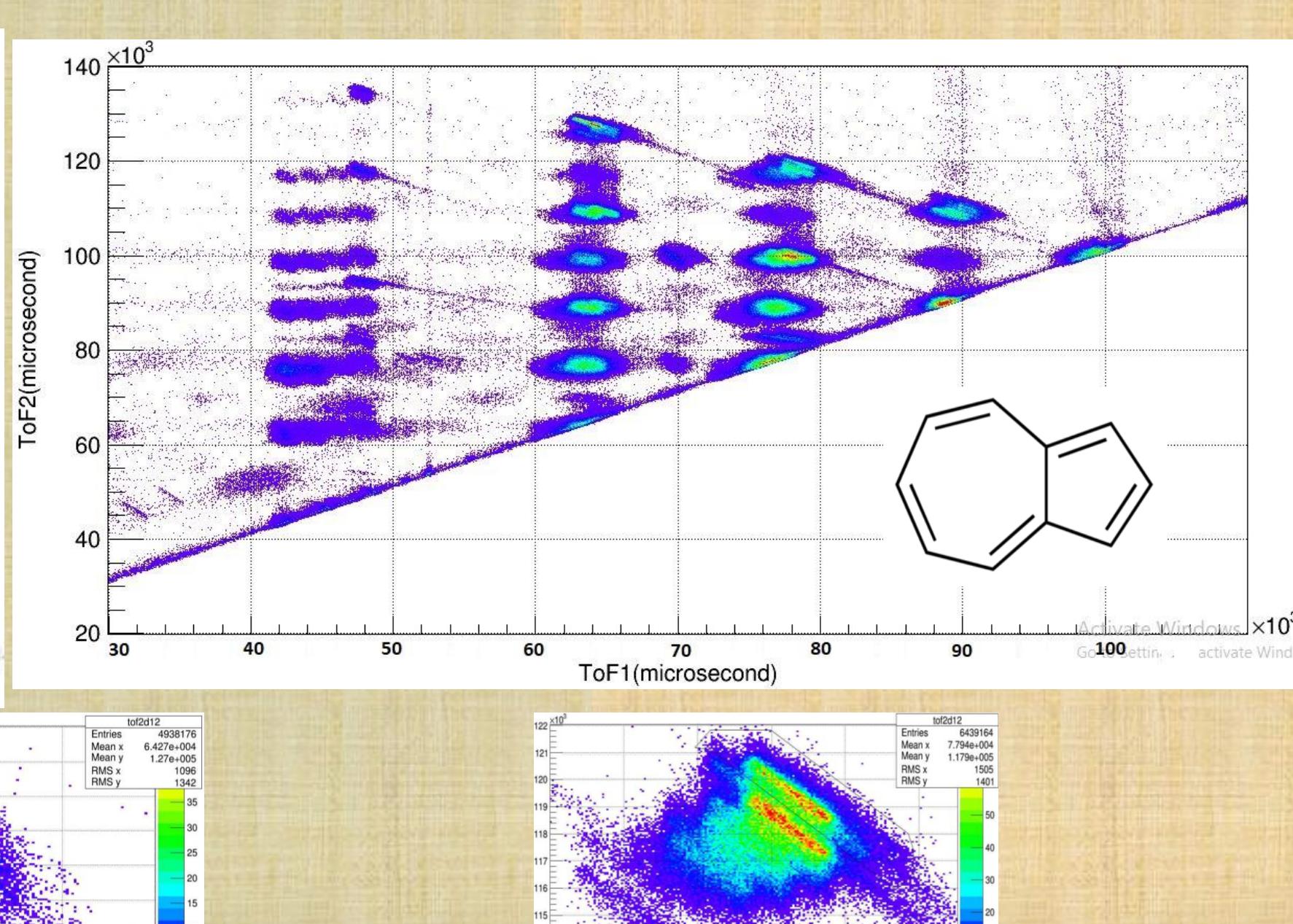
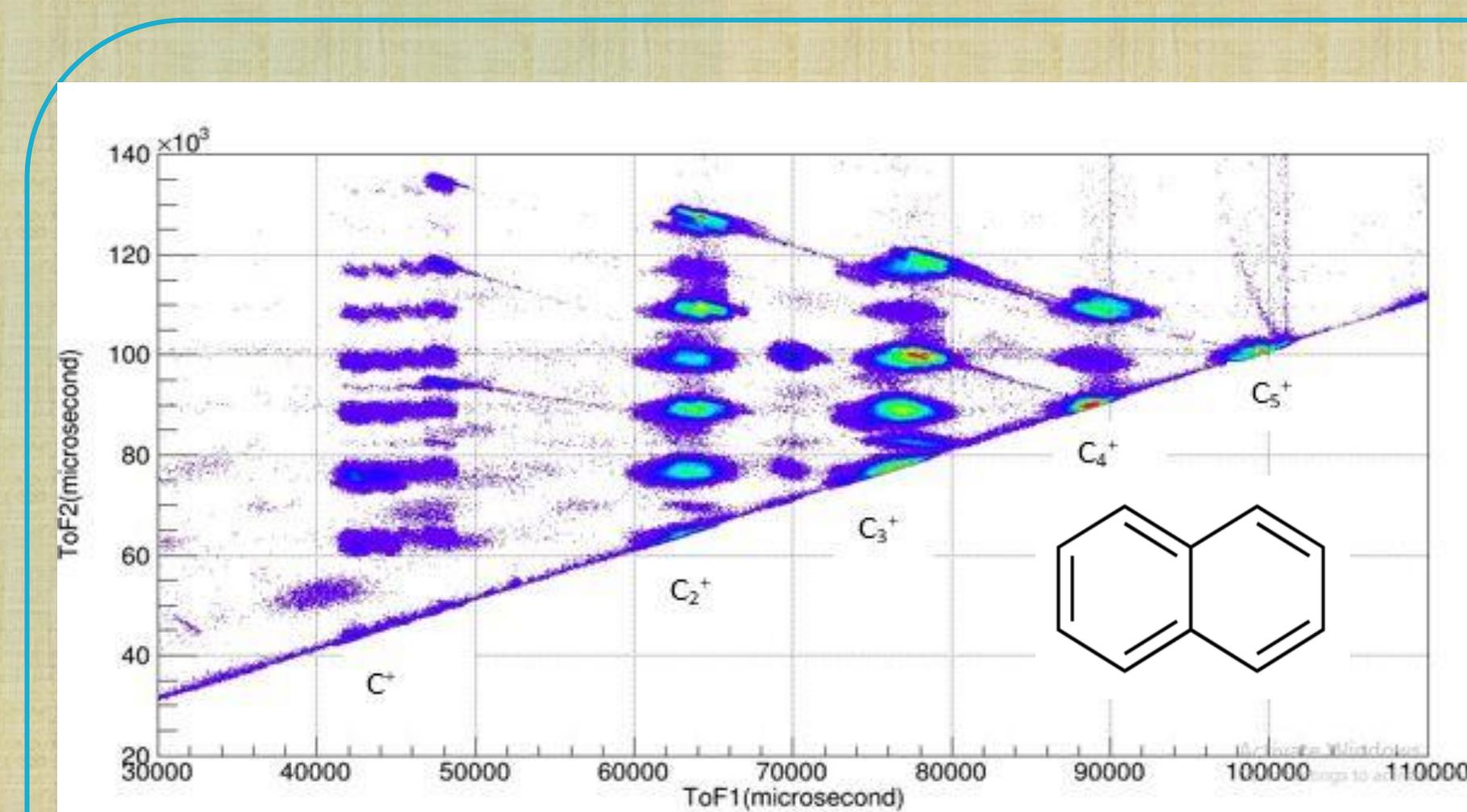
- For Nph^+ ion formed after plasmon excitation, the internal energy can be calculated by considering the plasmon energy – adiabatic IP.
- For Nph^+ ion formed after isomerisation of Az^+ , the internal energy can be calculated by considering the plasmon energy – adiabatic IP + isomerisation energy.



- Even the dication formation was observed to be sourced from double plasmon excitation.
- The relative cross sections of dication formation were well matched with double plasmon excitation model.

Suppression of H-loss channel seen in dication could also be addressed by double plasmon excitation. At the internal energy available of $\approx 13 \text{ eV}$, the main evaporation pathway is $2\text{H}/\text{H}_2$ loss channel. Whereas H loss dominates at lower internal energies (7–8 eV).

Ion-ion coincidence channels & KER values

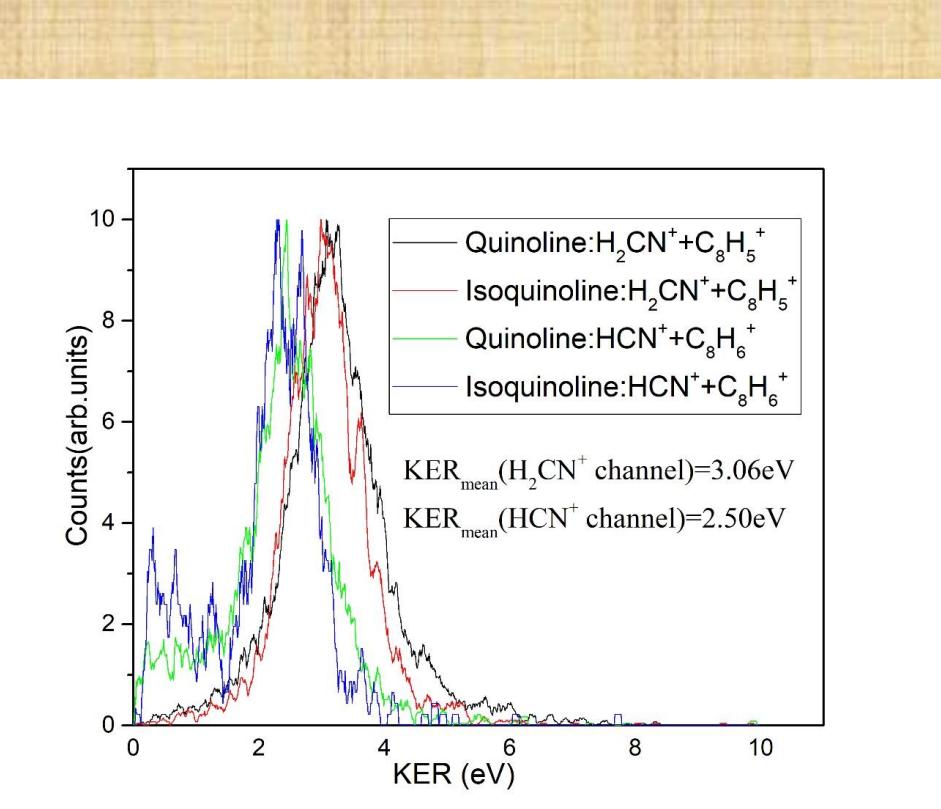
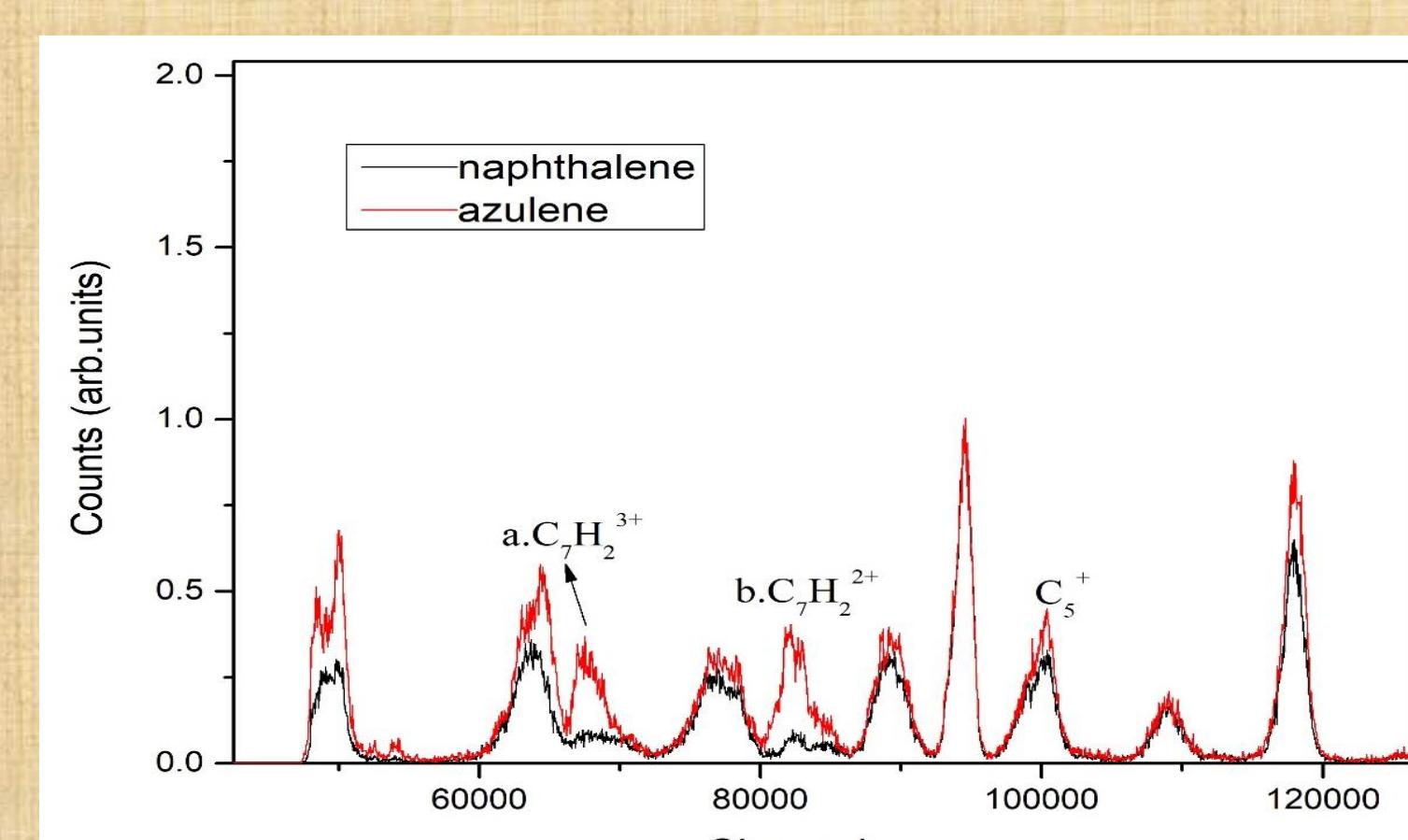


- A remarkable similarity is observed between KER spectra of azulene and naphthalene.
- It indicates both molecules have same dication precursors in double product reactions.
- KER is measured upto 3.26eV for carbon conserving channels.

References

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CH₃⁺ emission channels of azulene and naphthalene



KER spectra for PAH and PANH isomers for their binary fission channels

- A remarkable similarity is observed between ion-ion correlation spectra of isomeric PAHs & PANHs.
- Distribution of KER in the binary fission channels of isomeric PAHs and PANHs are measured to be identical.
- Identical KER spectra is obtained for quinoline and isoquinoline for the two binary channels.
- Unlike other channels, HCN⁺ emission has low KER component.
- KER distribution of H₂CN⁺ channel and C₂H₂⁺ emission channel of C₁₀H₈²⁺ are found to be identical.

Results & discussions

- KER is measured upto 3.26eV for carbon conserving channels.
- In contrast to naphthalene certain triply and quadruply charged azulene ion have higher propensity to decay via CH₃⁺ elimination.
- Emission of H₂CN⁺ from PANH dication is observed as one of the strongest channels among binary fission, corresponding molecule (amidogen) is a newly detected interstellar molecule.